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## A NEW SESQUITERPENE LACTONE GLUCOSIDE FROM *IXERIS SONCHIFOLIA*

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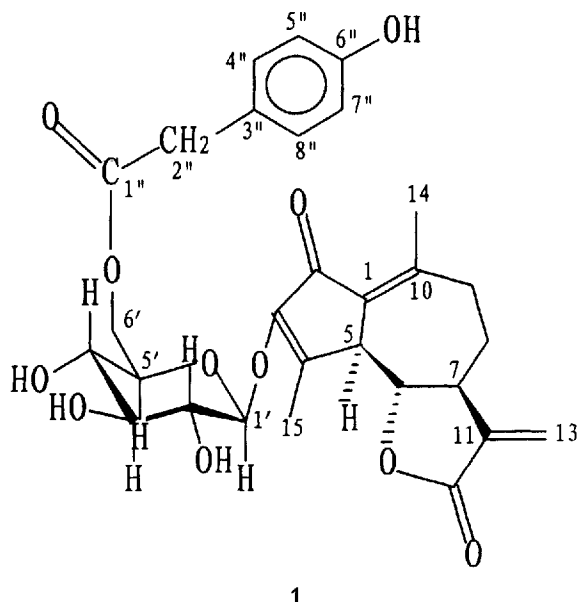
A new sesquiterpene lactone glucoside, Ixerin Z<sub>1</sub> (1), was isolated from the whole plants of *Ixeris sonchifolia* (Bge.) Hance, along with 15 known compounds. The structure of **1** was elucidated as 1(10),3,11(13)-guaiaatriene-12,6-olide-2-one-3-O-[6'-(*p*-hydroxyphenylacetyl)]-glucopyranoside by spectroscopic methods including 2D-NMR techniques.

**Keywords:** *Ixeris sonchifolia*; Compositae; Ixerin Z<sub>1</sub>

### INTRODUCTION

*Ixeris sonchifolia* (Bge.) Hance (Family Compositae) is abundantly distributed throughout the northeastern China. It is a small perennial herb, about 0.4 m tall commonly found in dry places. It has been used as a folk medicine in invigorating circulation of blood, normalizing menstruation and eliminating blood stasis to relieve pain [1]. Previous phytochemical studies on other species of this genus revealed the presence of sesquiterpene lactones [2–7]. In our investigations of biologically active and/or structurally novel substances from *I. sonchifolia*, we examined the constituents of the plant. This paper describes the isolation and structure determination of a new sesquiterpene lactone glucoside, Ixerin Z<sub>1</sub> (1).

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## RESULTS AND DISCUSSION

The methanol soluble fraction yielded Ixerin Z<sub>1</sub> (**1**), which was crystallized as white needles. A molecular formula of C<sub>29</sub>H<sub>32</sub>O<sub>11</sub> for compound **1** was assigned from its ESI-MS (M<sup>+</sup> + Na = 579) and EI-MS [M<sup>+</sup> + 1-(C<sub>6</sub>H<sub>11</sub>O<sub>5</sub> + C<sub>8</sub>H<sub>6</sub>O<sub>2</sub>) = 260]. The IR spectrum of **1** showed the presence of hydroxy groups (3400 cm<sup>-1</sup>), broad carbonyl bands at 1769 and 1670 cm<sup>-1</sup> indicating the presence of at least two different types of carbonyl groups, double bond (1640 cm<sup>-1</sup>) and aromatic ring (1616, 1517 cm<sup>-1</sup>). On acid hydrolysis, compound **1** gave glucose by comparison with authentic samples on co-TLC and paper chromatography. The <sup>1</sup>H-NMR spectrum (Tab. I) contained the characteristic signals located at δ 6.14 (1H, *d*, *J* = 3.0 Hz, H-13a) and δ 5.33 (1H, *d*, *J* = 3.0 Hz, H-13b) of an α-methylene-γ-lactone moiety. Signals for two vinyl methyls at δ 2.32 (3H, brs, H-15) and δ 2.47 (3H, brs, H-14) and a sugar moiety were also observed. A double doublet at δ 3.26 (1H, *J* = 10.2 Hz, 12.2 Hz) was assigned to H-6, which was coupled with H-5 at δ 3.30 (1H, *d*, *J* = 10.2 Hz) and H-7 at δ 2.75 (1H, brt, *J* = 12.2 Hz). This indicated the *trans*-diaxial relationship of the vicinal protons. The stereochemistry of these protons was considered to be H-5 α, H-6 β, since H-7 in naturally occurring guaianolides has α-orientation [8].

TABLE I  $^1\text{H-NMR}$  spectral data of **1** in pyridine- $d_5$ 

<i>H</i>	<b>1</b>	<i>H</i>	<b>1</b>
H-5	3.30 <i>d</i> ( $J = 10.2$ Hz)	H-2'	4.23 <i>dd</i> ( $J = 7.5, 8.0$ Hz)
H-6	3.26 <i>t</i> ( $J = 10.2, 12.2$ Hz)	H-3'	4.29 <i>dd</i> ( $J = 8.0, 8.0$ Hz)
H-7	2.75 <i>brt</i> ( $J = 12.2$ Hz)	H-4'	4.13 <i>dd</i> ( $J = 8.0, 8.0$ Hz)
H-8a	1.87 <i>brd</i> ( $J = 11.1$ Hz)	H-5'	4.10 m
H-8b	1.08 <i>dd</i> ( $J = 12.2, 11.1$ Hz)	H-6'a	4.99 <i>brd</i> ( $J = 11.4$ Hz)
H-9a	2.28 overlapped	H-6'b	4.73 <i>dd</i> ( $J = 11.4, 6.0$ Hz)
H-9b	2.08 m	H-2''	3.69 s
H-13	6.14 <i>d</i> ( $J = 3.0$ Hz)	H-4''	7.30 <i>d</i> ( $J = 8.4$ Hz)
	5.33 <i>d</i> ( $J = 3.0$ Hz)	H-5''	7.10 <i>d</i> ( $J = 8.4$ Hz)
H-14	2.47 <i>brs</i>	H-7''	7.10 <i>d</i> ( $J = 8.4$ Hz)
H-15	2.32 <i>brs</i>	H-8''	7.30 <i>d</i> ( $J = 8.4$ Hz)
H-1'	6.12 <i>d</i> ( $J = 7.5$ Hz)		

TABLE II  $^{13}\text{C-NMR}$  spectral data of **1** in pyridine- $d_5$ 

<i>Carbon</i>	$^{13}\text{C-NMR}$	<i>DEPT</i>	<i>HMBC</i>	<i>Carbon</i>	$^{13}\text{C-NMR}$	<i>DEPT</i>	<i>HMBC</i>
1	152.6	C	5,9,14	1'	101.4	CH	2',5'
2	188.5	C		2'	74.8	CH	1',3',4'
3	153.2	C	1',5,15	3'	77.6	CH	4',5'
4	146.6	C	5,15	4'	70.7	CH	3',5',6'
5	47.5	CH	6,15	5'	75.1	CH	1',3',4',6'
6	84.6	CH	5,7,8	6'	64.1	CH <sub>2</sub>	4',5'
7	51.9	CH	6,8,13	1''	171.6	C	6',2''
8	23.6	CH <sub>2</sub>	6,7,9	2''	39.7	CH <sub>2</sub>	4'',8''
9	36.4	CH <sub>2</sub>	8,14	3''	124.6	C	2'',5'',7''
10	129.1	C	5,9,14	4''	130.5	CH	2'',5'',7''
11	139.1	C	13	5''	115.8	CH	4'',8''
12	168.6	C	13	6''	157.3	C	4'',5'',7'',8''
13	117.5	CH <sub>2</sub>	9	7''	115.8	CH	4'',8''
14	21.2	CH <sub>3</sub>	5	8''	130.5	CH	2'',5'',7''
15	14.4	CH <sub>3</sub>	5				

The signals located at  $\delta$  7.30 (2H, *d*,  $J = 8.4$  Hz) and 7.10 (2H, *d*,  $J = 8.4$  Hz) showed the presence of a 1,4-disubstituted aromatic ring. The  $^{13}\text{C-NMR}$  data (Tab. II) of **1** showed the presence of 29 carbons. Of the fifteen unsaturated carbons, three were attributed to a lactone carbonyl at  $\delta$  168.6 (C-12), an  $\alpha$ ,  $\beta$ -unsaturated ketone carbonyl at  $\delta$  188.5 (C-2) and an ester carbonyl at 171.6 (C-1''), six were olefinic carbons and six were aromatic carbons; of the 14 alkyl carbons, seven were oxygen-bearing carbons of which six were derived from the glucose moiety. The remaining signals were those of two methines, three methylenes and two methyls. In the HMBC spectrum of compound **1**, the cross peaks between C-1'' with H-6'a and H-6'b confirmed that *p*-hydroxyphenylacetic acid was esterified at C-6 of the glucose moiety (Tab. II). The HMBC spectrum also displayed cross peaks between C-3 with H-1', showing that the glucose moiety was affixed to C-3

of the aglycone. The absolute configuration of the glucose moiety could not be deduced from the NMR data. The anomeric configuration was determined to be  $\beta$  from the  $J_{H1'-H2'}$  value ( $J=7.5$  Hz) [9]. Based on the above evidence, **1** was identified as 1(10),3,11(13)-guaiaatriene-12,6-olide-2-one-3-O-[6'-(*p*-hydroxyphenylacetyl)]-glucopyranoside.

## EXPERIMENTAL SECTION

### General Experimental Procedures

Instrumentation:  $^1\text{H-NMR}$  (300 MHz, pyridine- $d_5$ ) and  $^{13}\text{C-NMR}$  (75 MHz, pyridine- $d_5$ ), and 2D-NMR spectra were recorded on a Bruker AM 300 FT-NMR spectrometer using tetramethylsilane (TMS) as internal standard. Electrospray Ionization Mass (ESI-MS) and EI-MS were recorded on Finnigan LCQ LC/ESI-MS and VG 7070E spectrometer. IR spectrum was taken on a Bruker IFS 55 spectrometer and recorded in KBr pellets. Column chromatography was performed on silica gel (200~300 mesh, Qingdao, China), and the TLC analyses were carried out using glass precoated silica gel plates.

### Plant Material

The plant material was gathered in Liaoning Province, China, in June, 1996, and a voucher specimen (No.10082), identified by Prof. Qi-Shi Sun, is deposited in the Herbarium of the Department of Chinese Traditional Medicines, Shenyang Pharmaceutical University.

### Extraction and Isolation

The material was shade-dried and after grinding, 7.5 Kg were extracted with hot 70% EtOH three times. After removal of the solvent by evaporation, the EtOH extract (1500 g) was extracted three times each with petroleum ether (60~90°C),  $\text{CHCl}_3$ , and MeOH, under reflux. The MeOH extract was concentrated to a syrup (150 g), and then subjected to chromatographic separation on a silica gel column (200~300 mesh, 1400 g). The compounds of the mixture were eluted with  $\text{CHCl}_3$  (4 L) and with  $\text{CHCl}_3$ : MeOH (50:1, 30:1, 20:1, 12:1, 10:1, 9:1, 8:2, 7:3, 6:4, 5:5, each 10 L), subsequently 10 fractions were obtained. Rechromatography (60H silica gel, Qingdao, China) of the three fractions (6 g) [ $\text{CHCl}_3$ : EtOAc (7:3, v/v) as eluent] gave compound **1** (20 mg).

*Ixerin Z*<sub>1</sub> C<sub>29</sub>H<sub>32</sub>O<sub>11</sub>. Fine white needles. IR  $\nu_{\max}^{\text{KBr}}$  cm<sup>-1</sup>: 3440, 1770, 1670, 1652, 1620, 1517, 1446, 1385, 1253, 1070; positive ion ESI-MS m/z: 579 [M+Na]<sup>+</sup>, 557 [M+1]<sup>+</sup>, 260 [M+1-(C<sub>6</sub>H<sub>11</sub>O<sub>5</sub>+C<sub>8</sub>H<sub>6</sub>O<sub>2</sub>)]<sup>+</sup>; EI-MS (probe) 70ev, m/z (rel.int): 260 [M+1-(C<sub>6</sub>H<sub>11</sub>O<sub>5</sub>+C<sub>8</sub>H<sub>6</sub>O<sub>2</sub>)]<sup>+</sup> (7), 242 (2), 189 (3), 152 (17), 134 (12), 107 (100); <sup>1</sup>H and <sup>13</sup>C-NMR data are listed in Tables I and II.

*Acid hydrolysis of 1* Compound **1** was refluxed in 2.0 mol/L HCl at 100°C for 2 h. After cooling to room temperature, the reaction mixture was neutralized with Ag<sub>2</sub>CO<sub>3</sub> and centrifuged, and then the supernatant was evaporated on a water bath and subjected to TLC analysis on GF<sub>254</sub> [using CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O (6:4:1)] and paper chromatography [using *n*-BuOH-HOAc-H<sub>2</sub>O (4:1:5)] by comparison with authentic samples.

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